

## Mechanism of Low Temperature ALD of Al<sub>2</sub>O<sub>3</sub> on Graphene Terraces

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Uniform and defect-free Al<sub>2</sub>O<sub>3</sub> films were grown on highly oriented pyrolytic graphite (HOPG) terraces by thermal atomic layer deposition (ALD) at low temperature (50 °C) without any functionalization of the surface. Controlling the pulse times of trimethylaluminum (TMA) and H<sub>2</sub>O while using short Ar purge times, 1-2 nanometer size Al<sub>2</sub>O<sub>3</sub> particles were formed on the HOPG terraces. The particles provided a layer of nanoscale nucleation centers on the HOPG terraces. Capacitance-voltage measurements of Al<sub>2</sub>O<sub>3</sub> films grown at 50 °C using 50 ALD cycles showed an areal capacitance 1.17 μF/cm<sup>2</sup> with very small frequency dispersion, consistent with the absence of induction cycles and formation of a high quality interface. The leakage current of the Al<sub>2</sub>O<sub>3</sub> was ~10<sup>-5</sup> A/cm<sup>2</sup> in large area devices (1900 μm<sup>2</sup>) which is comparable with results from devices prepared using an identical ALD process for Al<sub>2</sub>O<sub>3</sub> on Si<sub>0.7</sub>Ge<sub>0.3</sub>(001) substrates, and is consistent with the absence of pinholes.

### Introduction

Since its emergence in 2004 (1,2), single layer graphene has attracted attention due to its high mobility of charge carriers and charge confinement. Graphene has been viewed as a viable alternative channel material for future logic transistors because transistors based on silicon are approaching their fundamental limits.(3,4) However, in order to fabricate graphene devices, deposition of ultrathin, pinhole-free high-k dielectric gates is required. Atomic layer deposition (ALD) is the most effective method to deposit ultrathin high-k dielectric layers, due to excellent thickness control.(5) However, because of the inert nature of graphene surfaces, similar to the surfaces of carbon nanotubes (CNTs), conventional ALD dielectric layers selectively nucleate on defect sites or step edges.(6,7) In the conventional ALD process on graphene, such non-uniform oxides result in large leakage currents in devices.(8)

For CNTs, Farmer et al functionalized the surface of CNTs using nitrogen dioxide (NO<sub>2</sub>) and trimethylaluminum (TMA) at 25 °C to form an Al<sub>2</sub>O<sub>3</sub> layer by ALD; however, the induction period for onset of uniform film growth was 100 ALD cycles.(9) Similarly,

several methods have been used to functionalize graphene surfaces including chemical treatment, deposition of oxidizing metal films, and polymer based seeding layers.(10, 11, 12, 13) Unfortunately, these functionalization methods can result in degradation of the electronic properties of graphene or require thick dielectric layers. Therefore, for successful fabrication of graphene devices, improved techniques for deposition of uniform and insulating gate oxides continue to be needed.

In this work,  $\text{Al}_2\text{O}_3$  was directly deposited on highly oriented pyrolytic graphite (HOPG) surfaces by low temperature thermal ALD using TMA and  $\text{H}_2\text{O}$  without any seeding layer or surface treatments prior to deposition. With the substrates at a temperature of  $50\text{ }^\circ\text{C}$ , using short purge times between the two precursor pulses and long pulses of the precursors, a CVD growth component was intentionally employed to provide more nucleation sites on the surface. The CVD growth component induces formation of 1-2 nanometer  $\text{Al}_2\text{O}_3$  particles on the surface which provide nucleation centers for uniform deposition. The deposited  $\text{Al}_2\text{O}_3$  film was continuous and uniform without observable defects. The surface morphology of the oxide was analyzed by atomic force microscopy (AFM) and electrical properties were characterized by capacitance-voltage and leakage current measurements of metal/ $\text{Al}_2\text{O}_3$ /HOPG stacks.

### Experimental Techniques

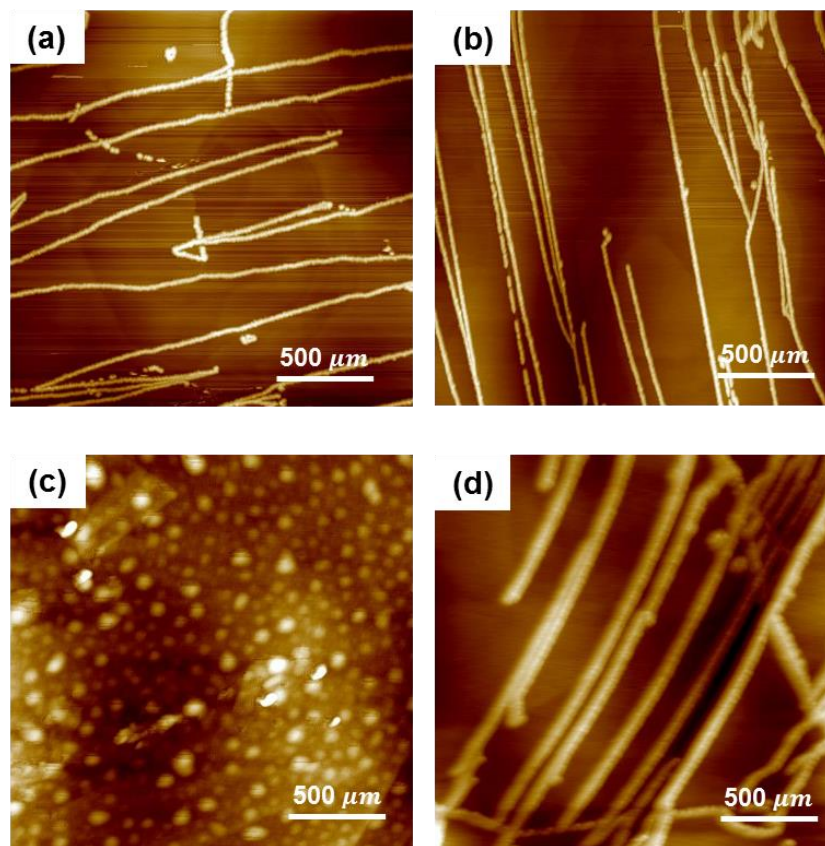
HOPG samples (from SPI supplies) were cleaned by the mechanical exfoliation method using an adhesive tape. The samples were loaded into a commercial ALD reactor (Beneq TFS 200 ALD system) which has a hot wall, crossflow reaction chamber. The reaction chamber was pumped down to 1 mTorr. To deposit  $\text{Al}_2\text{O}_3$ , TMA and  $\text{H}_2\text{O}$  were used as ALD precursors, temperature stabilized at  $20\text{ }^\circ\text{C}$ . The carrier gas was research purity argon (Ar, Praxair, 99.9999%) which was flowed at 300 sccm (standard cubic centimeter). 50 cycles of ALD were employed, each cycle consisting of a sequence of 600 ms TMA pulse, 500 ms Ar purge, 50 ms  $\text{H}_2\text{O}$  pulse, and 500 ms Ar purge. For comparison, using identical ALD pulse times, films were grown with sample temperatures between  $50\text{ }^\circ\text{C}$  to  $200\text{ }^\circ\text{C}$  to investigate the effect of temperature on the nucleation of  $\text{Al}_2\text{O}_3$ . To compare nucleation behavior on a highly reactive substrate,  $\text{Al}_2\text{O}_3$  was deposited on  $\text{Si}_{0.7}\text{Ge}_{0.3}(001)$  substrates using 50 cycles of ALD, consisting of sequences of 300 ms or 400 ms TMA pulses followed by 50 ms  $\text{H}_2\text{O}$  pulses with 500 ms Ar purge at  $50\text{ }^\circ\text{C}$  sample temperature. In addition, the effect of the precursor pulse times on the morphology of the oxide on HOPG were studied. TMA and  $\text{H}_2\text{O}$  pulse durations were changed from 200 ms to 600 ms and 50 ms to 150 ms, respectively, while the sample temperature during ALD was fixed at  $50\text{ }^\circ\text{C}$ .

After the ALD process, non-contact mode AFM measurements were performed to determine the correlation between the surface topography and the ALD conditions. MIM (Metal-Insulator-Metal) capacitors were fabricated to characterize the electrical properties of the oxide. In order to compare the quality of the oxide,  $\text{Si}_{0.7}\text{Ge}_{0.3}$  substrates were loaded along with the HOPG samples.  $\text{Al}_2\text{O}_3$  was deposited on HOPG and  $\text{Si}_{0.7}\text{Ge}_{0.3}(001)$  using 50 cycles of ALD with the samples at  $50\text{ }^\circ\text{C}$ . After the ALD process, Ni gates were deposited on the oxide by thermal evaporation. The gates were  $50\text{ }\mu\text{m}$  in diameter and 3 nm thick. As a control, Ni/ $\text{Al}_2\text{O}_3$ / $\text{Si}_{0.7}\text{Ge}_{0.3}$ /Al metal-oxide-

semiconductor capacitors (MOSCAPs) were also fabricated with a slightly different process due to different cleaning and contact requirements. Prior to ALD, each  $\text{Si}_{0.7}\text{Ge}_{0.3}(001)$  sample was treated with a 30 s rinse by each of acetone, isopropyl alcohol, and DI water followed by  $\text{N}_2$  drying. Afterwards, the native oxide was removed by cyclic HF cleaning using a 2% HF solution and DI water at 25 °C for 1 min in each solution for 2.5 cycles.(14) For each SiGe sample, 50 cycles of ALD deposition were followed by Ni gate deposition and 100-nm thick Al back contact deposition using DC sputtering. For both the HOPG and SiGe samples, the capacitance-voltage curves were measured in the frequency range of 2 kHz to 1MHz at room temperature with an HP4284A LCR meter. Leakage current of current of the oxide was obtained in the range of -2V to 2V.

## Results and Discussion

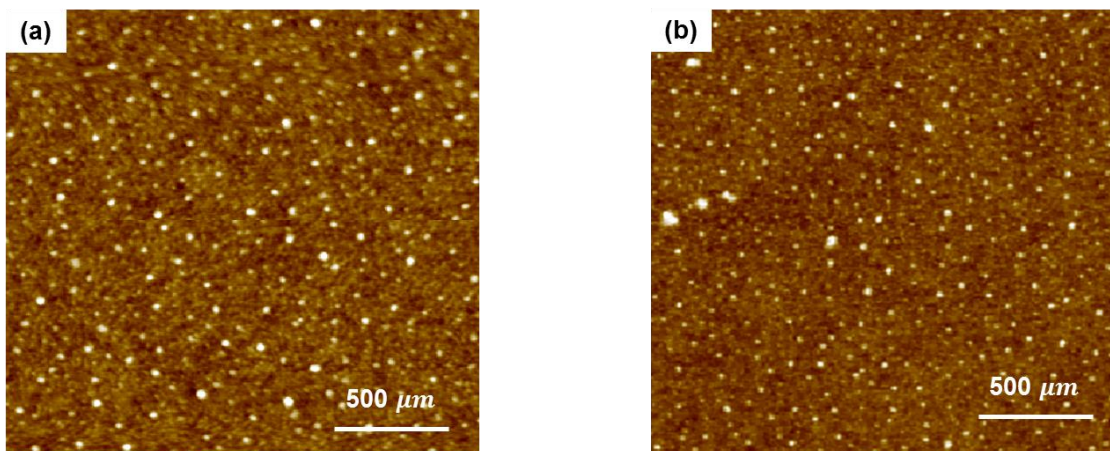
### Temperature dependence of $\text{Al}_2\text{O}_3$ ALD



**Figure 1. (Color) AFM images of  $\text{Al}_2\text{O}_3$  films (50 ALD cycles) on HOPG with different sample growth temperatures. (a) 200 °C, (b) 150 °C and (c) 50 °C. Each ALD cycle consisted of a 600 ms TMA pulse, 500 ms Ar purge, 50 ms  $\text{H}_2\text{O}$  pulse, and 500 ms Ar purge. (d) 50 ALD cycles of  $\text{Al}_2\text{O}_3$  were deposited with 2 s purge times at sample temperature of 50 °C, using otherwise identical ALD conditions as for the samples in (a), (b), (c). Non-contact mode AFM measurements were carried out with a Si tip. Note that for high temperature ALD, the nucleation of  $\text{Al}_2\text{O}_3$  was only observed along the step edges while for low temperature ALD, the nucleation was observed across the terraces and the step edges. In addition, for low temperature ALD, 1-2nm spheres were observed on the surface. The size of each image is  $2 \times 2 \mu\text{m}^2$ .**

Figure 1 shows AFM images of  $\text{Al}_2\text{O}_3$  grown on HOPG using 50 cycles of ALD at sample temperatures of 200 °C, 100 °C, and 50 °C. Each ALD cycle for the samples of Figs. 1(a),(b),(c) consisted of 600 ms TMA pulse and 50 ms  $\text{H}_2\text{O}$  pulse, with 500 ms Ar purge times between precursor pulses. The growth of  $\text{Al}_2\text{O}_3$  exhibited strong dependence on sample temperature. As shown in Figs. 1 (a) and (b), when the sample temperature was above 100 °C,  $\text{Al}_2\text{O}_3$  was only deposited on the step edges of the HOPG and not on the terraces, because dangling bonds for nucleation are only available on the step edges and not on the inert terraces. The thickness of the  $\text{Al}_2\text{O}_3$  deposited on the step edges was about 5 nm which is consistent with the expected thickness for 50 cycles of ALD at the typical 0.1 nm/cycle ALD growth rate.(15)

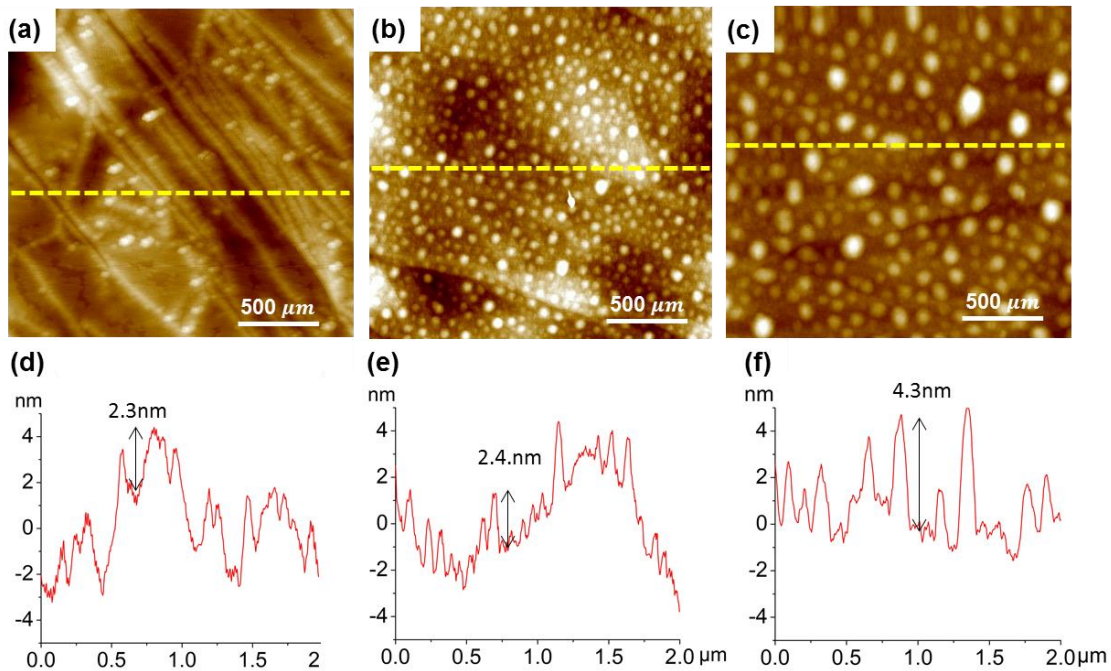
When the sample temperature was decreased to 50 °C as shown in Fig. 1(c), the  $\text{Al}_2\text{O}_3$  film was deposited continuously on both the step edges and the terraces without pin hole formation. Spherical  $\text{Al}_2\text{O}_3$  features were observed across the entire surfaces for the growth at 50 °C sample temperature. However, these features are only observed with a short purge time between two precursor pulses at 50 °C sample temperature. When the purge time was increased to 2s with identical precursor ALD pulse times and sample temperature, the spherical  $\text{Al}_2\text{O}_3$  features were not observed as shown in Fig 1(d). Short purge times can induce a chemical vapor deposition (CVD) growth component since excess unreacted physisorbed precursor molecules have insufficient time to desorb from both the substrate and chamber walls during the short purge time. The CVD component induced deposition of particles of about  $2 \pm 0.4$  nm diameter which was quantified by averaging 10 AFM line scans.  $\text{Al}_2\text{O}_3$  particles on the surface were formed by either nucleation in the ALD chamber and subsequently transported to the sample surface or were formed by precursor islands on the surface. Since surface island growth is usually non-spherical and similar nuclei were also observed on the reactive SiGe surface (Fig 2(a),(b)), the data is most consistent with nuclei formation in the ALD chamber (denoted gas phase nucleation below).(16, 17)



**Figure 2. (Color) AFM images of  $\text{Al}_2\text{O}_3$  films (50 ALD cycles) on  $\text{Si}_{0.7}\text{Ge}_{0.3}(001)$  grown at 50 °C sample temperature. ALD was carried out using sequences of (a) 300 ms TMA pulse and 50 ms  $\text{H}_2\text{O}$  pulse with 500 ms of Ar purge after each precursor pulse, (b) 400 ms TMA pulse and 50 ms  $\text{H}_2\text{O}$  pulse with 500 ms of Ar purge after each precursor pulse at 50 °C sample temperature. Spherical  $\text{Al}_2\text{O}_3$  nuclei were observed on SiGe (white dots). This data is consistent with nuclei formation in the ALD chamber. The size of each image is  $2 \times 2 \mu\text{m}^2$ .**

## Effect of precursor pulse time on Al<sub>2</sub>O<sub>3</sub> ALD

Figure 3 (a)-(c) shows the AFM images of Al<sub>2</sub>O<sub>3</sub> grown by ALD at 50 °C sample temperature with different pulse lengths of TMA and H<sub>2</sub>O with a fixed Ar purge time of 500 ms. For the sample with 200 ms TMA pulses and 50 ms H<sub>2</sub>O pulses (Fig. 3(a)), Al<sub>2</sub>O<sub>3</sub> was mainly deposited on the step edges. Although some Al<sub>2</sub>O<sub>3</sub> was nucleated on the terraces, it was discontinuous with a high density of visible pinholes. The number densities of the Al<sub>2</sub>O<sub>3</sub> particles (number of particles per 4 μm<sup>2</sup> image area) with the three different ALD conditions are shown in Table I. For the sample shown in Fig. 3(a), grown using relatively short 200 ms TMA pulses, the density of the Al<sub>2</sub>O<sub>3</sub> particles (15/μm<sup>2</sup>) was significantly lower than for the samples shown in Figs. 3(b),(c), which were grown under different conditions with longer TMA pulse times. When the TMA pulse time was increased to 600 ms while fixing the H<sub>2</sub>O pulse length (Fig. 3(b)), the density of the Al<sub>2</sub>O<sub>3</sub> particles was markedly increased (123/μm<sup>2</sup>) and continuous Al<sub>2</sub>O<sub>3</sub> films were deposited on both terraces and step edges without pinholes. AFM line traces show that the particles are 2 ± 0.6 nm in diameter (Fig 3(e)). The observation of the high density of defects and low density of Al<sub>2</sub>O<sub>3</sub> particles in the dielectric deposited using short TMA pulses indicates that the Al<sub>2</sub>O<sub>3</sub> particles have a critical role in the formation of uniform dielectric layers on HOPG during ALD.



**Figure 3. (Color) AFM images of Al<sub>2</sub>O<sub>3</sub> films (50 ALD cycles) on HOPG** (a) 200 ms TMA and 50 ms H<sub>2</sub>O pulses, (b) 600 ms TMA and 50 ms H<sub>2</sub>O pulses, and (c) 600 ms TMA and 150 ms H<sub>2</sub>O pulses. (d), (e), and (f) are AFM height profiles along the yellow lines in Figs. (a), (b) and (c) respectively. For the ALD with short pulses ((a), (d)), few condensation nuclei were observed, and the ALD nucleation primarily occurred on the step edges. For the two samples grown with longer TMA pulses ((b), (e) and (c), (f)), white condensation nuclei were distributed across the terraces, and ALD nucleation occurred on both the terraces and the step edges.

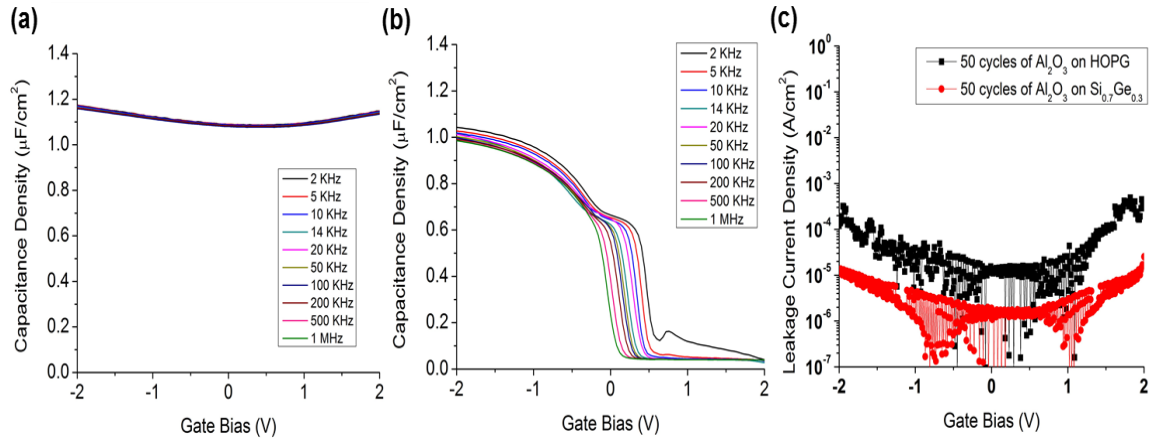
When the H<sub>2</sub>O pulse time was increased to 150 ms with a fixed TMA pulse time of 200 ms (Fig. 3(c)), similar morphology as for the growth with a long TMA pulse (Fig. 3(b)) was observed. The Al<sub>2</sub>O<sub>3</sub> film was continuous with a high density of Al<sub>2</sub>O<sub>3</sub> particles (44/μm<sup>2</sup>). AFM lines traces (Fig. 3(f)) show the particles increased in size to 4 ± 0.7 nm. This indicates that the Al<sub>2</sub>O<sub>3</sub> particles were formed by a CVD component that can be controlled by the TMA and H<sub>2</sub>O pulse times. This observation is consistent with both the island formation mechanism and the gas phase formation mechanism for the particles.

**TABLE I. Number density of Al<sub>2</sub>O<sub>3</sub> particles from films grown with different ALD conditions (Number of particles per 4 μm<sup>2</sup>)**

200 ms TMA & 50 ms H <sub>2</sub> O pulses	600 ms TMA & 50 ms H <sub>2</sub> O pulses	200 ms TMA & 150 ms H <sub>2</sub> O pulses
59	692	176

### Electrical Properties

MIM capacitors were fabricated on freshly cleaved HOPG substrates. Al<sub>2</sub>O<sub>3</sub> films were deposited using 50 ALD cycles consisting of 600ms TMA pulse, 500 ms Ar purge, 50 ms H<sub>2</sub>O pulse, and 500 ms Ar purge at 50 °C sample temperature. Subsequently, Ni metal gates were deposited by thermal evaporation. The area of the capacitor was ~1900 μm<sup>2</sup> (50 μm diameter). Capacitance-voltage (C-V) and leakage current-voltage (I-V) measurements were performed in order to evaluate the electrical quality of the oxide. As shown in Fig. 4(a), the capacitance of the oxide was nearly independent of the applied voltage owing to the MIM structure of the capacitor. Previously, Park et al reported fully modulated capacitance of Al<sub>2</sub>O<sub>3</sub>/monolayer TiOPc/graphene stacks.(18) For single layer graphene, capacitance can be modulated near 0 V due to the linear dispersion of the density of states near the Fermi energy level. However, because of the high charge carrier density of HOPG near the Fermi energy level, the modulation of capacitance was not observed. The C<sub>max</sub> of the oxide was 1.17 μF/cm<sup>2</sup> which is consistent with the C<sub>max</sub> value of the ALD grown Al<sub>2</sub>O<sub>3</sub> on Si<sub>0.7</sub>Ge<sub>0.3</sub>(001) as shown in Fig. 4(b). This indicates that deposition of Al<sub>2</sub>O<sub>3</sub> on HOPG proceeded without an ALD induction time. Note the high dispersion of the C-V data on Si<sub>0.7</sub>Ge<sub>0.3</sub> was likely due to the formation of GeO<sub>x</sub> which can be suppressed by (NH<sub>4</sub>)<sub>2</sub>S(aq) or NH<sub>3</sub> plasma treatments.(19, 20) However, these cleaning recipes were not employed in this study since they may induce unintentional chemical changes of HOPG and degradation of the surface quality on HOPG.



**Figure 4. (Color) Comparison of Electrical Properties of Low Temperature Al<sub>2</sub>O<sub>3</sub> ALD on HOPG and Si<sub>0.7</sub>Ge<sub>0.3</sub>(001) with cycles consisting of a 600 ms TMA pulse, 500 ms Ar purge, 50 ms H<sub>2</sub>O pulse, and 500 ms Ar purge at 50 °C sample temperature.**(a) Capacitance vs. Voltage Curve of Ni/Al<sub>2</sub>O<sub>3</sub>(50 ALD cycles)/HOPG stack. (b) Capacitance vs. Voltage Curve of Ni/ Al<sub>2</sub>O<sub>3</sub>(50 ALD cycles)/Si<sub>0.7</sub>Ge<sub>0.3</sub>/Al stack. (c) I-V curve of Ni/ Al<sub>2</sub>O<sub>3</sub>(50 ALD cycles)/HOPG and Ni/ Al<sub>2</sub>O<sub>3</sub>(50 ALD cycles)/Si<sub>0.7</sub>Ge<sub>0.3</sub>/Al stack. The similar Cox on HOPG and Si<sub>0.7</sub>Ge<sub>0.3</sub>(001) is consistent with no ALD induction cycles on HOPG. The comparable leakage currents of Al<sub>2</sub>O<sub>3</sub>/HOPG to Al<sub>2</sub>O<sub>3</sub>/Si<sub>0.7</sub>Ge<sub>0.3</sub> are consistent with the oxide on HOPG being uniform and pin-hole free on the HOPG substrate.

Figure 4 (c) compares the leakage currents of Al<sub>2</sub>O<sub>3</sub> films grown by 50 ALD cycles on HOPG (red) and Si<sub>0.7</sub>Ge<sub>0.3</sub>(001) (blue) substrates. ALD was performed simultaneously at 50 °C sample temperature on the two substrates. The leakage current of the oxide on the HOPG was  $3.1 \times 10^{-5}$  A/cm<sup>2</sup> and  $2.2 \times 10^{-6}$  A/cm<sup>2</sup> for the Si<sub>0.7</sub>Ge<sub>0.3</sub> at -1 V. The low leakage on Si<sub>0.7</sub>Ge<sub>0.3</sub> is expected since Al<sub>2</sub>O<sub>3</sub> readily nucleates on Si<sub>0.7</sub>Ge<sub>0.3</sub>(001), and the Si<sub>0.7</sub>Ge<sub>0.3</sub>(001) surface is flat without bunched steps. Conversely, HOPG surfaces are inert and have bunched steps.(16, 18) The leakage current of the ALD grown Al<sub>2</sub>O<sub>3</sub> on HOPG at 50 °C sample temperature being within 15x of the leakage current on Si<sub>0.7</sub>Ge<sub>0.3</sub>(001) is consistent with the oxide on HOPG being uniform and pin-hole free on the HOPG substrate.

## Conclusion

Deposition of high quality Al<sub>2</sub>O<sub>3</sub> films on HOPG was demonstrated by low temperature ALD without evidence for an ALD induction period prior to onset of uniform film growth. Controlling the pulse times of TMA and H<sub>2</sub>O along with a short purge time, 1-2 nanometer diameter spherical Al<sub>2</sub>O<sub>3</sub> particles were formed on both HOPG and SiGe consistent with a gas phase reaction of the ALD precursors. The nuclei provided uniform nucleation centers on the inert HOPG surface resulting in uniform and pin-hole free Al<sub>2</sub>O<sub>3</sub> films on both step edges and terraces. Comparable C<sub>ox</sub> for low temperature ALD grown Al<sub>2</sub>O<sub>3</sub> on HOPG and Si<sub>0.7</sub>Ge<sub>0.3</sub>(001) is consistent with the absence of induction cycles even on the inert HOPG surface. The leakage current of the Al<sub>2</sub>O<sub>3</sub> was as low  $\sim 10^{-5}$  A/cm<sup>2</sup> which is within 15x of the leakage current observed for

Al<sub>2</sub>O<sub>3</sub> on highly reactive Si<sub>0.7</sub>Ge<sub>0.3</sub>(001) substrates. This work has great potential for fabrication of novel graphene-based devices.

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