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Passivation of InGaAs(001)-(2x4) by Self-limiting CVD of a Silicon Hydride Control Layer

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ABSTRACT:A saturated Si-Hx seed layer for gate oxide or contact conductor ALD has been deposited via two separate self-limiting and saturating CVD processes on InGaAs(001)-(2x4) at substrate temperatures of 250° C and 350° C. For the first self-limiting process, a single silicon precursor, Si_3H_8 , was dosed at a substrate temperature of 250°C and XPS results show the deposited silicon hydride layer saturated at about 4 monolayers of silicon coverage with hydrogen termination. STS results show the surface Fermi level remains unpinned following the deposition of the saturated silicon hydride layer, indicating the InGaAs surface dangling bonds are electrically passivated by Si-Hx. For the second self-limiting process, Si_2Cl_6 was dosed at a substrate temperature of 350°C and XPS results show the deposited silicon chloride layer saturated at about 2.5 monolayers of silicon coverage with chlorine termination. Atomic hydrogen produced by a thermal gas cracker was subsequently dosed at 350°C to remove the Si-Cl termination by replacing with Si-H termination as confirmed by XPS, and STS results confirm the saturated Si-H_x bilayer leaves the InGaAs(001)-(2x4) surface Fermi level unpinned. Density function theory modeling of silicon hydride surface passivation shows an Si-Hx monolayer can remove all the dangling bonds and leave a charge balanced surface on InGaAs.

INTRODUCTION

InGaAs intrinsically has very high electron mobility, making it a leading material for replacement of silicon as the n-type channel of metal oxide semiconductor field effect transistors (MOSFETs).1-3 Subnanometer FET devices have lower threshold voltages thereby making trap and fixed charge elimination critical.⁴⁻⁷ The III-V(001) surfaces are dominated by relatively unreactive filled and empty dangling bonds while the group IV dangling bonds are half filled and, therefore, highly reactive. The high reactivity of the group IV dangling bonds facilitates their elimination during deposition of gates oxides. Therefore, it is proposed that a thin layer (2-4 monolayers) of epitaxial crystalline silicon on InGaAs or related materials (GaAs or InAs) could facilitate dangling bond elimination and formation of a passive interface.

Previous work has shown the deposition of a thin molecular beam epitaxy (MBE) silicon interfacial control layer on the clean $In_{0.53}Ga_{0.47}As(100)$ surface prior to SiO₂ deposition serves to reduce surface Fermi level pinning,^{7,8} faces eliminates ACS Paragon Plus Environment

Deposition of a thin MBE silicon layer on In_{0.53}Ga_{0.47}As(100) prior to Al₂O₃ deposition lead to improved MOSFET device performance by decreasing frequency dispersion, reducing hysteresis, and lowering D_{it}.⁸ Similarly, previous reports also show the physical vapor deposition (PVD) of an a-silicon passivation layer prior to deposition of a bilayer gate insulator stack (1 nm Al₂O₃/5 nm HfO₂) passivates In_{0.53}Ga_{0.47}As, as C-V characteristics showed inversion and low D_{it}, and XPS results show no III-V oxide formation.⁹ Ex-situ deposition of a thin \sim 1.2 nm a-silicon layer by PECVD on In_{0.53}Ga_{0.47}As prior to Al₂O₂ gate oxide ALD has been shown to suppress gallium suboxide formation and MOSFET devices exhibit higher drive current and higher effective electron mobility values.^{10,11} Detailed XPS studies have shown the presence of Ga2O3 at the interface of both GaAs and In0.53Ga0.47As based MOSFET devices leads to high frequency dispersion and consequently high Dit, while an ex-situ deposition (PECVD) of an a-silicon layer on GaAs/In_{0.53}Ga_{0.47}As surfaces eliminates Ga +3 oxide formation and significantly

lowers frequency dispersion while increasing device drive current in MOSFET C-V measurements.¹²

Si based interface control layers have also been investigated on GaAs(100) based MOSFET devices by PVD and PECVD techniques. Physical vapor deposition (PVD) of multilayer silicon has been used to passivate GaAs(100), but this method has been shown to require a minimum thickness of 1.5 nm in order to effectively passivate the III-V surface and protect against oxygen diffusion to the III-V surface and the concomitant formation of high D_{it} during high temperature annealing.^{13,14} GaAs(100) based MOSFET devices with a 1.2 nm a-silicon passivation layer deposited by ex-situ plasma enhanced chemical vapor deposition (PECVD) prior to Al₂O₃ gate oxide ALD have been fabricated and show low frequency dispersion, negligible false inversion, and high drive current as the silicon layer getters oxygen, protecting from higher III-V oxidation states.15,16

The silicon MBE and PVD processes leave the surface terminated with unpassivated silicon atoms which have half-filled dangling bonds which pin the Fermi level and readily react with trace gasses such as H₂O. Chemical vapor deposition (CVD) processes expose the substrate to one or more gaseous precursors in order to grow conformal thin films at a given temperature. CVD growth is usually not limited to a single atomic layer, as heterogeneous and homogeneous surface reactions may occur. In an atomic layer deposition (ALD) process, precursors and growth temperature are chosen to inhibit gas phase reactions and to limit chemical reactivity to heterogeneous surface species. An ALD reaction always employs two precursors, one an oxidant and one a reductant, which are dosed in separate "half reactions" to avoid homogeneous reactions. Surface dangling bonds are terminated at each ALD half-cycle reaction; since one reactant is an oxidant while the other is a reductant, each heterogeneous halfcycle is self-limiting. At the end of the ALD process, the surface is terminated with one of the ALD precursors and usually no dangling bonds. Consequently, ALD allows deposition of silicon with dangling bonds already passivated by ALD precursor ligands thereby providing chemical passivation.

Commercial ALD flow-type reactors operate at 0.75 -7.5 torr where one ALD cycle consists of a pulse of the first ALD precursor, followed by an inert gas purge to remove traces of unreacted precursor and gaseous byproducts, followed by a pulse of the second ALD precursor and a subsequent inert gas purge.¹⁷ Many Surf Sci reports include self-limiting and saturating atomic layer deposition half-cycle reactions in a high vacuum (base pressure <10^-3 torr), or ultra-high vacuum environment (base pressure <~10^-9 torr) and will be referred to here as HV-ALD or UHV-ALD. The high/ultra-high vacuum environment aides in protecting the substrate and deposited thin films from unwanted contaminants which may be more prevalent in higher base pressure systems. In HV-ALD or UHV-ALD, the inert gas purge steps are usually replaced

by simple vacuum purge steps. Previously, a silicon HV-ALD growth process was reported on SiO₂ using alternate pulses of Si₂H₆ and SiCl₄ at substrate temperatures of 355-385°C, and each ALD cycle required several minutes as formation of the HCl(g) byproduct is slow below 400°C.¹⁸ Silicon UHV-ALD was reported on Ge substrates by alternating pulses of Si₂Cl₆ and atomic hydrogenor by alternating pulses of SiH₂Cl₂ and SiH₄ at substrate temperatures of 400° - 560°C.^{19,20} Silicon UHV-ALD processes on Si substrates have been reported with alternating pulses of SiH₂Cl₂ and atomic hydrogen or by alternating pulses Si_2H_6 and Si_2Cl_6 , at substrate temperatures (400° -560°C).^{21,22} These HV/UHV-ALD processes employ high substrate temperatures which probably desorb the passivating ligand and may cause changes in substrate reconstruction or composition for InGaAs and related materials.23

In this report, a self-limiting and saturating HV-CVD process using a single silicon ALD precursor (Si₃H₈) at a low substrate temperature of 250°C is compared with a second self-limiting and saturating HV-CVD growth process employing the silicon precursor, Si₂Cl₆, at a substrate temperature of 350°C. The Si₃H₈ process results in depositing a saturated thin silicon hydride capping layer (approximately 4 monolayers) on the InGaAs(001)-(2x4) surface and leaving the surface Fermi level unpinned and ready for subsequent atomic layer deposition (ALD) deposition of the gate oxide. Once the Si-H groups passivate substrate dangling bonds, saturation occurs because H, desorption from silicon dihydride is slow at 250°C and H₂ desorption from silicon monohydride species does not occur below 445°C.²⁴ The Si₂Cl₆ self limiting CVD process results in a thin saturated silicon bilayer deposited on the InGaAs(001)-(2x4) surface and the InGaAs surface is left terminated with Si-Cl groups. Atomic hydrogen produced by a thermal gas cracker is subsequently dosed at 350°C to quickly remove the Si-Cl termination by replacing with Si-H termination through a ligand exchange reaction resulting in the HCl(g) desorption byproduct. This process also leaves the surface Fermi level unpinned and ready for subsequent surface functionalization with an oxidant or further silicon multilayer growth by ALD.As far as we know, this is the first report of epitaxial deposition of 2-4 layers of silicon with hydrogen termination by selflimiting CVD; the process is inherently self-limiting because it takes advantage of the low desorption temperature of hydrogen and chlorine from InGaAs relative to silicon.

EXPERIMENTAL DETAILS

This study employs n-type (Si dopant) and p-type (Be dopant) samples consisting of 0.2 μ m of 1-2x10¹⁸ doped In_{0.53}Ga_{0.47}As layers grown by MBE on commercially available InP substrates. The samples were capped with a 50nm As² layer and shipped/stored under vacuum prior to being loaded into the Omicron ultrahigh vacuum (UHV) preparation chamber with a base pressure of 1x10⁻¹⁰ Torr. In the preparation chamber, the samples were degassed at

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58 59 60 250°C for 30 minutes and, subsequently, decapped and annealed for one hour by radiatively heating at 360-370°C to obtain the InGaAs(001)-(2x4) surface reconstruction.

Following annealing, the samples were characterized by an Omicron in situ monochromatic XPS using the aluminum K α excitation source (hv=1486.7 eV) with spectra taken at a glancing angle of 30° to obtain enhanced surface sensitivity. XPS raw counts were collected using the XPS constant analyzer energy mode with a pass energy of 50eV and line width of 0.1 eV. XPS peak shape analysis was conducted using CASA XPS v.2.3 by employing a Shirley background subtraction. All XPS raw core level peaks were corrected by Schofield photoionization cross sectional relative sensitivity factors. Oxygen contamination was monitored over the course of the experiments by XPS and percentages were calculated by dividing the O is corrected peak area by the sum of the As 2p, Ga 2p, and In 3d, and Si 2p corrected peak areas. For the Si₃H₈ process, the percent O was below the XPS detection limit and for the Si₂Cl₆ process, the percent O was <5% [see supplemental material]. Carbon contamination for both processes was below the XPS detection limit. Following XPS elemental analysis of the surface, the samples were transferred to the SPM analysis chamber which has a base pressure of 2x10⁻¹¹ Torr. In the SPM chamber, scanning tunneling microscopy (STM) was performed at 300 K to determine the atomic order of the surface by using constant current mode with the tunneling current set point at 0.1 nA and the sample bias set to -3 V for filled state Scanning tunneling spectroscopy (STS) was imaging. performed to determine the electrical quality of the surface and probe the local surface density of states using variable-z mode with the sample bias swept from -1.5 to +1.5 V and the tip simultaneously moving towards and then away from the surface.²⁵ An applied Δz initial offset ranging from -0.2 to -0.8 nm was used in order to maximize I(V) signal without crashing the STM tip. The dI/dV spectra were recorded using a lock-in amplifier and STS curves are reported by averaging 10-12 single curves taken across the sample surface.

After initial characterization, the samples were transferred back to the preparation chamber and radiatively heated to 250° C or 350° C for 15 minutes while simultaneously the high vacuum ALD chamber manipulator was also heated to 250° C or 350° C for 15 minutes to facilitate a faster sample transfer. Prior to sample transfer to the HV-ALD chamber, Si₃H₈ or Si₂Cl₆ were prepulsed in the ALD dosing chamber to coat the chamber walls with precursor prior to sample transfer. This preheating and prepulsing procedure was done before every dose of Si₃H₈ or Si₂Cl₆.

Both processes are referred to as CVD processes because the InGaAs substrate undergoes some surface induced etching by hydrogen or chlorine dissociated ligands from Si₃H₈or Si₂Cl₆, keeping these processes from classification as true ALD.^{26,27} The HV-CVD processes include the substrate in the HV-ALD chamber (base pres-

sure $<2x10^{-7}$ Torr) undergoing exposure to Si₂H₈ or Si₂Cl₆ at 250°C or 350°C with the exposure measured in Langmuirs (1x10⁻⁶ Torr/1 second) by a convectron gauge located adjacent to the HV-ALD chamber. The HV-CVD process transfers well into a commercial ALD tool as demonstrated in the Beneq TFS-200 continuous flow reactor where MOSCAP fabrication employing the Si₂Cl₆ based passivation process was demonstrated [see supplemental material]. For the saturating Si₂H₈ based process, the sample was transferred into the HV-ALD dosing chamber and 13 MegaLangmuir Si₃H₈ was dosed at 250°C. Following Si₃H₈ dosing, the sample was transferred back to the preparation chamber where XPS was performed. Following an initial 13 MegaLangmuir Si_2H_8 dose, the sample was exposed to a series of additional doses at 250°C to increase the integrated dose to 50, 100, and 300 MegaLangmuir of Si₃H₈. The 13, 50, and 100 MegaLangmuir total Si_2H_8 doses consisted of 10 second pulses of 7.5 x 10⁻² Torr and the 300 MegaLangmuir total dose (additional 200 MegaLangmuir) consisting of 10 second pulses of 1x10⁻¹ Torr. After each dose, the sample was transferred to the preparation chamber for XPS studies. After the 300 MegaLangmuir dose, the sample was transferred to the SPM chamber for STM and STS. Following STM and STS of the complete 300 MegaLangmuir dosed surface, the sample was subsequently annealed to 450°C for 30 minutes at a heating rate of 2 K/s. During annealing, a mass spectroscopy measurement was taken with a quadropole mass spectrometer equipped with an RGA detector located in the preparation chamber (SRS RGA100). The RGA detector was operated in histogram mode to examine the partial pressure versus mass of detected gas species by sweeping across o-70 amu. STS was performed to determine any effect on the surface Fermi level position following high temperature annealing. Mass Spectroscopy shows 2x higher presence of H2 species (m=2 amu) seen during the 30 minute 450°C anneal [see supplemental material].

For the self-limiting and saturating CVD growth process with Si₂Cl₆, the sample was transferred into the ALD dosing chamber and 3 MegaLangmuir Si₂Cl₆ was dosed at 350°C. Following the initial 3 MegaLangmuir Si₂Cl₆ dose, the sample was exposed to a series of additional doses at 350°C to increase the integrated dose to 12, and 21 Mega-Langmuir of Si₂Cl₆. After each Si₂Cl₆ dose, the sample was transferred back to the preparation chamber where XPS was performed. The 3, 12, and 21 MegaLangmuir total Si_2Cl_6 doses consist of 10 second pulses of 2.5 x 10⁻² Torr. After the complete 21 MegaLangmuir dose, the sample was dosed with 500 Langmuirs of atomic hydrogen in order to remove the surface chlorine termination by replacing with hydrogen termination through a ligand exchange reaction creating an HCl(g) byproduct. An Oxford Applied Research TC-50 thermal gas cracker was employed to produce atomic hydrogen and was operated at 65 Watts, producing atomic hydrogen at 50% efficiency. The 500 Langmuir dose consists of $H_2(g)$ flowed for 8 minutes and 20 seconds at an H_2 pressure of 1×10^{-6} Torr; note the calculated Langmuirs does not include the H cracking fraction since this could not be experimentally verified so the reported atomic H Langmuirs are an upper limit.

The Density-Functional Theory (DFT) simulations were performed using VASP plane-wave DFT simulation package with projector augmented-wave (PAW) pseudopotentials²⁸⁻³³ and PBE exchange-correlation functional.³⁴ The InGaAs was a regular polymorph with 50% Ga and 50% In. Initially, the InGaAs unitcell was optimized at variable volume to avoid internal compression/strain. The optimized unitcell was later used to build the InGaAs supercell and initial slabs with desired surfaces. All slab relaxations were performed using conjugate-gradient relaxation algorithm with a force tolerance level of 0.05 eV/Å and Gamma-centered 5x7x1 K-point grid. During relaxations the 3 bottom layers of InGaAs slabs were permanently fixed in their bulk-like positions and saturated with pseudo-H atoms with 1.25 |e| charge to simulate continuous bulk. The Si atoms at upper surfaces were passivated by normal 1.0 |e| H atoms. To avoid possible dipoles, dipole correction in vertical Z direction was applied.

RESULTS

Figure 1(a) shows the raw XPS peak areas for Ga 3p and Si 2p peaks on the clean (2x4) surface, and following 13, 50, 100, and 300 MegaLangmuir Si₃H₈ doses. The Ga 3p spin orbit split peaks are located at binding energies 104.4 eV and 107.9 eV, and the Si 2p peak is located at a binding energy of 99.5 eV. With each additional dose, the Si 2p peak area increases and the substrate Ga 3p peak decreases, indicative of increasingly higher surface coverage of SiH_x groups. Figure 1(b) shows the corrected XPS peak areas for the clean n-type InGaAs(001)-(2x4) surface, and following 13, 50, 100, and 300 MegaLangmuir Si₃H₈ doses. In 3d, Ga 2p, and As 2p higher binding energy peaks were chosen in order to analyze the top most monolayers of the surface. The corrected peak area of In 3d is about three times higher than Ga 2p and As 2p corrected peak areas on the clean decapped surface consistent with the In 3d having a binding energy less than half the binding energy of Ga 2p and As 2p (therefore a greater probe depth), as well as the reported phenomenon that indium segregates to the top most surface layers of InGaAs(001) alloys.³⁵⁻³⁷ The decrease in the In 3d, Ga 2p, and As 2p substrate peaks with an increase in silicon coverage is consistent with a uniform surface coverage of silicon. The Si 2p corrected peak area following the 13 MegaLangmuir Si₃H₈ dose is half that of the total In, Ga, and As combined substrate peak areas. The total 300 MegaLangmuir Si₃H₈ dose is 23 times the initial 13 MegaLangmuir dose yet the increase in silicon coverage is only 2.7. The XPS data is consistent with a self-limiting CVD growth process.

To further quantify the saturation, Fig. 1 (c) shows change in silicon coverage versus Si_3H_8 dose for 13, 50, 100, and 300 MegaLangmuir total doses on n-type In-

GaAs(001)-(2x4) at 250°C. The change in silicon coverage was calculated for each Si_3H_8 dose by dividing the increase in corrected Si 2p peak area by the increase in dose.



Figure 1.XPS of Si₃**H**₈ **dosed InGaAs(001)-(2x4)**.XPS raw counts are corrected by Schofield photoionization cross sectional relative sensitivity factors. (a) Raw XPS peak areas for Ga 3p and Si 2p on clean InGaAs(2x4), and following 13, 50, 100, and 300 MegaLangmuir total Si₃H₈ doses at 250°C. (b) XPS corrected peak areas for 13, 50, 100, and 300 MegaLangmuir total Si₃H₈ doses at 250°C on n-type InGaAs(001)-(2x4). (c) Rate of change of silicon coverage versus total Si₃H₈ dose as calculated from XPS corrected Si 2p peak areas for 13, 50, 100, and 300 MegaLangmuir total Si₃H₈ doses at 250°C on n-type InGaAs(001)-(2x4). (d) XPS corrected peak area ratio of Si/(In+Ga+As) for clean (2x4), 13 Mega Langmuir Si₃H₈, 50 Mega Langmuir Si₃H₈ total doses at 250°C. Note error bars are standard errors.

The thickness of the deposited silicon capping layer can be calculated from the equation $\ln(I/I_0) = -t/\lambda$, where I is the sum of the intensity of the In 3d, Ga 2p, and As 2p peaks following each Si₃H₈ dose, I₀ is the sum of the intensity of the In 3d, Ga 2p, and As 2p peaks on the clean InGaAs(001)-(2x4) surface, t is the thickness of the deposited silicon layer, and λ is the inelastic mean free path of the collected electrons of the InGaAs substrate (1 nm). Using this equation, the total 300 MegaLangmuir Si₃H₈ dose saturates at about 4.5 monolayers of silicon coverage with hydrogen termination. The calculated silicon thickness closely corresponds with the ratios shown in Fig. 1 (d).

Figure 2 (a) shows a filled state STM image of the clean InGaAs(001)-(2x4) surface and (b) shows 300 MegaLangmuir Si₃H₈ dosed on the p-type InGaAs(001)-(2x4) surface at 250°C with no further annealing. Vertical ordering along the same direction as the underlying substrate arsenic dimer (2x4) rows is observed in regions across the

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STM image. Five line traces are taken across these ordered regions (Fig. 2(c)). Spacing between ordered rows is nearly identical to the clean (2x4) surface with average spacing at 1.5±0.26 nm consistent with III-V dangling bond elimination through silicon bonding in a commensurate structure with the substrate in regions across the surface. Line traces were measured across the surface as shown in fig. 2(d); surface features vary in height by one atomic step (~2.3±0.2 Å standard error) showing high surface uniformity. Previous STM studies of MBE silicon growth on the GaAs(001)-(2x4) surface show silicon absorbs in localized heteroepitaxial ordered structures across the (2x4) surface with several surface reconstructions present, consistent with our finding of local surface epitaxy.³⁸ As shown in previous work, epitaxial growth of silicon on the In_{0.53}Ga_{0.47}As surface is not required for surface passivation and improving MOSFET device performance.⁷⁻¹¹ This self-limiting CVD process using Si₂H₈ results in local surface epitaxy as shown with both STM and XPS measurements, as the Si 2p peak shows resolved spin-orbit splitting [see supplemental].39



Figure 2.STM/STS of Si₃**H**₈ **dosed InGaAs(001)-(2x4).** (a) Large area filled state STM image of the clean decapped In-GaAs(001)-(2x4) surface (b) enlarged area filled state STM images following 300 MegaLangmuir Si₃H₈ dose at 250°C on p-type InGaAs(001)-(2x4) with no further annealing. Dotted lines in (b) indicate rows in which line trace are taken across. Note some ordered rows are shifted 45°, consistent with a mixture of Si-H and Si-H₂ species terminating the surface. (c) Four line traces taken across ordered regions of the surface with average row spacing at 1.5 +/- 0.26 nm. Arrows point to the atomic positions in each line trace. (d) Line traces taken

across the filled state STM image shown in (b). Surface features vary in height by a maximum of one atomic step (~2.3 +/- o.2 Å). (e) STS results of the decapped n-type In-GaAs(oo1)-(2x4) surface, the 300 MegaLangmuir total Si₃H₈ dose at 250°C, and the 300 MegaLangmuirSi₃H₈ dosed surface following a 30 minute anneal at 450°C (average of 11 STS curves are shown). Note the surface Fermi level shifts towards mid-gap after annealing, indicative of surface dangling bonds created by increasing the rate ofdesorption of H₂ (g) from silicon. (f) STS results of the decapped p-type In-GaAs(001)-(2x4) surface before and after the 300 MegaLangmuir total Si₃H₈ dose at 250°C. The alignment of both the valence and conduction bands and the Fermi level position remains the same on both n-type and p-type surfaces following the saturation dose, indicative of surface passivation.

Figure 2 (e) shows the STS measurements probing the local surface density of states of the n-type decappedIn-GaAs(001)-(2x4) surface before and after a saturated 300 MegaLangmuir Si_3H_8 dose at 250°C. It has previously been shown that InGaAs(001)-(2x4) based MOSFETs exhibit good Fermi level modulation characteristics, indicating the InGaAs(001)-(2x4) surface Fermi level is unpinned.⁴⁰ The conduction and valence band edges align with that of the clean surface showing a saturation Si₂H₈ dose at 250°C leaves the surface Fermi level unpinned. Figure 2 (e) also shows subsequently annealing the saturated 300 MegaLangmuir Si₃H₈ on n-type InGaAs(001)-(2x4) surface to 450° C for 30 minutes causes the surface Fermi level to shift towards mid-gap, indicative of surface dangling bonds created by desorption of H_2 (g) from silicon due to the high temperature annealing. Similar results are seen on the p-type InGaAs(001)-(2x4) surface showing the surface Fermi level also remains unpinned and p-type following a saturation 300 MegaLangmuir Si₃H₈ dose at 250°C [Figure 2(f)]. Desorption limited CVD growth occurs at 250°C, where the presence of any available clean InGaAs surface sites provide a path for H₂ recombinative desorption to occur and further silicon multilayer growth to continue slowly until saturation is found at about 4.5 monolayers of silicon with hydrogen termination. Si-H₃ is the least stable hydride species with desorption on a silicon terminated surface occurring at 225°, leaving Si-H₂ and Si-H groups deposited on the surface which are stable up to 330°C.⁴¹ H₂ desorption is close to zero at 250°C on a silicon terminated surface consistent with the surface being saturated with Si-H, and Si-H species after all clean InGaAs surface sites have reacted with silicon.²⁴ The H₂ desorption peak from arsenic-rich surface reconstructions of GaAs(001) starts at 225°C.²⁶ The 300 MegaLangmuir total Si₃H₈ dose (additional 200 MegaLangmuir Si₃H₈ following the total 100 MegaLangmuir Si₂H₈ dose) takes over half an hour to complete at the indicated dosing pressure. This lengthy dose time assists in the slow H₂ desorption from remaining clean InGaAs surface sites, and from neighboring Si-H₂ sites, where H₂ recombinatively desorbs at low temperatures until nearly complete surface saturation with Si-H and Si- H_2 species is reached.²⁴

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Figure 3 (a) shows the raw XPS peak areas for Ga 3p and Si 2p peaks on the clean (2x4) surface, and following 3, 12, and 21MegaLangmuir Si₂Cl₆ doses. The Ga 3p spin orbit split peaks are located at binding energies 104.4 eV and 108.2 eV, the Si 2p spin orbit split peaks is located at binding energy 99.8 eV. The Si 2p spin orbit split peaks are well resolved for 12 and 21 MegaLangmuir Si₂Cl₆ total doses and a new Si 2p chemical state is seen at a higher binding energy (100.5-101 eV) indicative of Si-Cl_x bonding or Si-O_x bonding from the small amount of oxygen contamination seen over the course of the experiment [see supplemental material].42,43 With each additional dose, the Si 2p peak area increases and the substrate Ga 3p peak decreases, indicative of increasingly higher surface coverage of silicon until the surface reaches saturation and is left terminated by chlorine. Figure 3 (b) shows the surface composition from XPS for the clean n-type InGaAs(001)-(2x4) surface, and following 3, 12, and 21 MegaLangmuir Si₂Cl₆ doses. In 3d, Ga 2p, and As 2p higher binding energy peaks are chosen in order to analyze the top most monolayers of the surface. Note the decrease in the In 3d, Ga 2p, and As 2p substrate peaks with a concurrent increase in silicon coverage consistent with a uniform surface coverage of silicon. The Si 2p peak area following the 3 MegaLangmuir Si_2Cl_6 dose is about 5/6 that of the total In, Ga, and As combined substrate peak areas. The total 21 MegaLangmuir Si₂Cl₆ dose is 7 times the initial 3 Mega-Langmuir dose yet the increase in silicon coverage is only 1.87. The XPS data is consistent with a self-limiting CVD growth process. The XPS data in Figure 3 (b) shows that the chlorine signal is negligible following the initial Si₂Cl₆ dose but is more prominent following the 12 and 21 Mega-Langmuir doses consistent with excess surface gallium and indium being preferentially etched by chlorine following the initial 3 MegaLangmuir Si₂Cl₆ dose.²⁷ It is hypothesized that once excess surface gallium and indium has been etched and all clean In, Ga, and As surface sites have reacted with Si-Cl_x groups, the surface becomes saturated by chlorine termination as shown following the total 21 MegaLangmuir Si₂Cl₆ dose. Chlorine desorption from silicon is close to zero at 350°C.44

Fig. 3 (c) shows change in silicon coverage versus Si_2Cl_6 dose for 3, 12, and 21 MegaLangmuir total doses on n-type InGaAs(001)-(2x4) at 350°C as well as an additional 500 Langmuir atomic hydrogen dosed at 350°C following the total 21 MegaLangmuir Si_2Cl_6 dose at 350°C. The change in silicon coverage is calculated for each Si_2Cl_6 dose by dividing the increase in corrected Si 2p peak area by the increase in dose. Self-limiting and saturating coverage of silicon on the InGaAs(001)-(2x4) surface is seen following the saturated 21 MegaLangmuir Si_2Cl_6 dose as no further increase in the rate of silicon coverage is observed. Similar results are seen on the p-type InGaAs(001)-(2x4) surface. Figure 3(d) shows the Si/(In+Ga+As) XPS peak area ratios for clean n-type (2x4), 3 MegaLangmuir, 12 Mega-

Langmuir, and 21 MegaLangmuir total Si₂Cl₆ doses at 350°C before and after an additional 500 Langmuir atomic hydrogen dose at 350°C. The ratios shown in Fig. 3 (d) correspond with the thickness of the deposited silicon capping layer, which was calculated from the equation $\ln(I/I_0) = -t/\lambda$, where I is the sum of the intensity of the In 3d, Ga 2p, and As 2p peaks following each Si₂Cl₆ dose, I₀ is the sum of the intensity of the In 3d, Ga 2p, and As 2p peaks (001)-(2x4) surface, t is the thickness of the deposited silicon layer, and λ is the inelastic mean free path of the collected electrons of the In-GaAs substrate (1 nm). Using this equation, the total 21 MegaLangmuir Si₂Cl₆ dose saturates at about 2.5 monolayers of silicon coverage.



Figure 3.XPS of Si₂Cl₆ dosed InGaAs(001)-(2x4).XPS raw counts are corrected by Schofield photoionization cross sectional relative sensitivity factors. (a) Raw XPS peak areas for Ga 3p and Si 2p on clean InGaAs(2x4), and following 3, 12, and 21 MegaLangmuir total Si₂Cl₆ doses at 350°C.(b) XPS corrected peak areas for 3 MegaLangmuir Si₂Cl₆, 12 MegaLangmuir Si₂Cl₆, 21 MegaLangmuir Si₂Cl₆, and 21 MegaLangmuir Si₂Cl₆ + 500 Langmuir atomic hydrogenon ntype InGaAs(001)-(2x4). All doses done at 350°C. (c) Rate of change of silicon coverage versus total Si₂Cl₆dose as calculated from XPS corrected Si 2p peak areas for 3, 12, and 21 MegaLangmuir total Si₂Cl₆doses at 350° C on n-type InGaAs(001)-(2x4). (d) XPS corrected peak area ratio of Si/(In+Ga+As) for clean (2x4), and 3, 12, and 21 MegaLangmuir total Si₂Cl₆doses at 350°C on n-type InGaAs(001)-(2x4).

Figure 4 (a) (a) and (b) shows filled state STM images of 21 MegaLangmuir Si₂Cl₆ dosed on the n-type InGaAs(00)-(2x4) surface at 350°C followed by 500 Langmuir atomic hydrogen dosed at 350°C with no further annealing. Vertical ordering along the same direction as the underlying substrate arsenic dimer (2x4) rows is observed on regions of the surface and four line traces are taken across these

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58 59 60 ordered regions (Fig. 4(c)). The spacing between rows is nearly identical to the clean (2x4) surface with average spacing at 1.6±0.1 nm consistent with III-V dangling bond elimination through silicon bonding locally in a commensurate structure. Line traces were measured across the surface as shown in Fig. 4(d); surface features vary in height by one atomic step (~2.4 \pm 0.1 Å), showing high surface uniformity.

Figure 4 (e) shows the STS measurements of the n-type decapped InGaAs(001)-(2x4) surface before and after saturation Si₂Cl₆ dosing followed by 500 Langmuir atomic hydrogen, and 4 additional 9 MegaLangmuir Si₂Cl₆ + 500 Langmuir of atomic H ALD cycles all dosed at 350°C. The conduction and valence band edges align with that of the clean surface showing a saturation Si₂Cl₆ dose followed by atomic hydrogen dose at 350°C leaves the surface Fermi level unpinned. Similar results are seen on the p-type InGaAs(001)-(2x4) surface showing the surface Fermi level also remains unpinned and p-type following saturation Si₂Cl₆ dosing, 500 Langmuir atomic hydrogen, and following an additional 4 Si₂Cl₆ + atomic H ALD cycles all dosed at 350°C [Figure 4 (f)].



Figure 4.STM/STS of Si₂Cl₆ **dosed InGaAs(001)-(2x4).** (a) Large area and (b) enlarged area filled state STM images following 21 MegaLangmuir Si₂Cl₆ dose at 350° C and 500 Langmuir atomic hydrogen dosed at 350° C on n-type

InGaAs(001)-(2x4) with no further annealing.Dotted lines indicate rows in which line trace are taken across.Arrows point to atomic positions in each line trace. Note some ordered rows are shifted 45°, consistent with a mixture of Si-H and Si-H₂ species terminating the surface. (c) Four line traces taken across ordered regions of the surface with average row spacing at 1.6 +/- 0.1 nm. (d) Line traces taken across the filled state STM image shown in (a). Surface features vary in height by one atomic step ($\sim 2.4 + / - 0.1 \text{ Å}$) showing high surface uniformity. (e) STS on n-type InGaAs: clean n-type InGaAs (2x4), 3 MegaLangmuir dose of Si_2Cl_6 at 350°C, 21 MegaLangmuir Si₂Cl₆, 500 L atomic Hydrogen, and 4 ALD cycles at 350°C (an average of 9 STS curves are shown). Each Silicon ALD cycle = 9 MegaLangmuir Si_2Cl_6 at 350°C followed by 500 Langmuir atomic H at 350°C. (f) STS on p-type InGaAs: clean n-type InGaAs (2x4), 3 MegaLangmuir dose of Si₂Cl₆ at 350°C, 21 MegaLangmuir Si₂Cl₆and 500 Langmuir of atomic hydrogen, and an additional 4 ALD cycles at 350°C (an average of 9-13 STS curves are shown). Note the alignment of both the valence and conduction bands following the saturation dose as well as the Fermi level position remaining the same on both ntype and p-type surfaces indicative of surface passivation.

DFT simulations of the initial stages of silicon hydride passivation of the InGaAs(001)-(2x4) surface are shown in Fig. 5. Fig. 5 (a) and (b) shows partial coverage (less than 1 monolayer) of Si-H₂ groups bonding to surface arsenics. Fig. 5 (c) and (d) shows full monolayer coverage of Si-H and Si-H₂ groups on the InGaAs surface containing As bulk-like termination with surface arsenics bonding to a mixture of Si-H and Si-H₂ groups. As seen in XPS results for the Si₂Cl₆ based process, the InGaAs surface is increasingly arsenic rich with each increasing dose of Si₂Cl₆. Following the initial 3 MegaLangmuir Si₂Cl₆ dose, the As 2p peak becomes asymmetric due to the formation of As-Si bonds, leading to a chemical shift of As 2p to a higher binding energy [see supplemental material]. The passivation model shown in Fig. 5 (e) and (f) contains a topmost InGaAs surface layer comprised of $\frac{1}{2}$ arsenic and $\frac{1}{2}$ indium/gallium atoms bonding to Si-H, consistent with the XPS data from the Si₃H₈ based process showing nearly equivalent amounts of gallium and arsenic on the surface. Following the initial 13 MegaLangmuir Si₂H₈ dose, the As 2p and Ga 2p peaks become asymmetric due to the formation of As-Si and Ga-Si bonds, leading to a chemical shift of As 2p to a higher binding energy and a chemical shift of Ga 2p to a slightly lower binding energy [see supplemental material]. All initial DFT calculations of the silicon hydride passivation of the InGaAs(001)-(2x4) surface shown in Fig. 5 are in agreement with the STS results showing the surface Fermi level remaining unpinned.



Figure 5.Initial DFT simulations of silicon hydride passivation of InGaAs(001)-(2x4).(a) DFT model of partial coverage of Si-H₂ groups passivating the InGaAs(001)-(2x4) surface. (b) Calculated local density of states for the partial coverage DFT model shown in (a). Note the Fermi level remains unpinned following partial coverage passivation. (c) DFT model of full coverage of Si-H₂/Si-H groups passivating the InGaAs(001)-(2x4) surface. Note the topmost InGaAs surface is arsenic rich with one Si-H group and one Si-H₂ group bonding to each surface arsenic atom. (d) Calculated local density of states for the full coverage DFT model shown in (c). The Fermi level remains unpinned following full coverage passivation. (e) DFT model of full coverage of Si-H₂/Si-H groups passivating the InGaAs(001)-(2x4) surface. Note the topmost InGaAs surface contains 1/2 arsenic and 1/2 indium/gallium atoms bonding to Si-H_x groups. (f) Calculated local density of states for the full coverage DFT model shown in (e). The Fermi level remains unpinned following the passivation.

M.D. Pashley described the electron counting model applied to the (In)GaAs(001)-(2x4) reconstructed surface which stated surface bonding conditions necessary to maintain no net surface charge, a condition essential to creating an unpinned surface Fermi level.⁴⁵ L. Lin and J. Robertson employ this electron counting rule model to create surface interfacial passivating layers on reconstructed semiconductor surfaces by maintaining no net charge for each successive layer of growth.⁵ The electron counting rule model has been applied to all DFT models shown in Fig. 5 with number of valence electrons for indium/gallium, arsenic, silicon, and hydrogen are 3, 5, 4, and 1. For the partial coverage model shown in Fig. 5 (a),

the sp² hybridized indium/gallium surface atoms accommodate the charge deficiency found on each of the sp³ hybridized arsenic surface atoms. The full monolayer coverage model shown in Fig. 5 (c) is comprised of all sp³ hybridized atoms. Here each surface arsenic bonds to 1 Si-H group (containing 1 filled dangling bond) and 1 Si-H₂ group, leaving the overall unit cell charge balanced. The full coverage model shown in Fig. 5 (e) contains 4 sp² hybridized Si-H groups which contain excess electrons to balance the charge deficiency on the sp³ hybridized surface indium/gallium and arsenic atoms leaving the unit cell charge neutral.

CONCLUSION

Deposition of a thin silicon hydride capping layer on InGaAs(001)-(2x4) has been achieved via two separate self-limiting CVD processes as shown by XPS. The 250°C Si₃H₈ process only requires the use of a single ALD precursor, Si₂H₈, with self-limiting growth of 4.5 monolayers of Si-H_x coverage achieved at a very low temperature. The Si₂Cl₆350°C process produces a thinner Si-H_x capping layer (2.5 monolayers) and allows for multilayer silicon growth by ALD through cyclically dosing Si₂Cl₆ and atomic hydrogen. STM and STS measurements show both selflimiting CVD processes on InGaAs(001)-(2x4) produce an atomically locally ordered and electrically passivated surface, with the surface Fermi level (E_F) shifting from the valence to the conduction band for p-type vs. n-type samples consistent with an unpinned E_F. Initial DFT calculations show the InGaAs(001)-(2x4) surface is electronically passivated by Si-H_x groups via the satisfaction of the electron counting rule for a charge neutral interface. The calculated local density of states are in agreement with the experimental STS measurements showing the surface Fermi level remains unpinned, and the passivating silicon hydride control capping layer is ready for ALD gate oxide nucleation. Initial MOSCAP fabrication results also show the insertion of a silicon passivation layer by dosing Si₂Cl₆ on the InGaAs(001) surface prior to the deposition of Al_2O_3 leads to lower frequency dispersion, higher C_{max} , and a smaller false inversion indicative of lower D_{it} at midgap.46 The initial device results show the deposited silicon layer with hydrogen termination seeds high-K gate oxide nucleation, and improves device performance.

ASSOCIATED CONTENT

Supporting Information

Figures S1-S6.This material is available free of charge via the Internet at http://pubs.acs.org.

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